The Variational Nature of some Reaction Path Models.

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Abstract

The reaction path is a central subject in theoretical chemistry. It is a pathway imagined on the potential energy surface (PES). It provides a one-dimensional description of a chemical reaction in an N-dimensional configuration space. There are a variety of reaction path models. Each type of reaction path has advantages and inconvenients in the description of the process. We present the variational nature of the most widely used models, namely, Intrinsic Reaction Coordinate, Newton Trajectory, Gradient Extremal, and Gentles Ascent Dymanic *Path.* Extensions of the Reaction Path model is also discussed.

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1.- The Reaction Path Model.

The Reaction Path Model.



A comparison between the Reaction Path described by the Intrinsic Reaction Coordinate and a representative, long exact classical trajectories. From a physical point of view any Reaction Path (RP) can be seen as if one envisions a large cloud of classical trajectories evolving on the PES. Very often the average trajectory is going to be close to the curve selected for the RP. This is the physical meaning of the Reaction Path Model. (J. González, X. Giménez, J. M. Bofill, *Phys. Chem. Chem. Phys.* 4, 2921 (2002))

The Reaction Path Model.

The model of Reaction Path: R.A. Marcus, *J. Chem. Phys.* **49**, 2610, 2617 (1968) and K. Fukui, *J. Phys. Chem.* **74**, 4161 (1970).



Reaction Path: <u>A continuous curve monotonicaly increasing in energy from a minimum of the</u> <u>Potential Energy Surface to a First Order Saddle Point and monotonicaly decreasing from this</u> <u>point to a new minimum of this Potential Energy Surface.</u>

There is different type of curves satisfying the definition of Reaction Path.

The most widely used curves are:

The Steepest Descent Curve (Intrinsic Reaction Coordinate) K. Fukui, J. Phys. Chem. **74**, 4161 (1970)

Gradient Extremal

M. V. Basilevsky, A. G. Shamov, *Chem. Phys.* **60**, 347 (1981)

Newton Trajectory (Distinguished Reaction Coordinate)

M. J. Rothman, L. L. Lohr, *Chem. Phys. Lett.* **70**, 405 (1980);

W. Quapp, M. Hirsch, O. Imig, D. Heidrich, *J. Comput. Chem.* **19**, 1087 (1998); J. M. Anglada, E. Besalú, J. M. Bofill, R. Crehuet, *J. Comput. Chem.* **22**, 387 (2001)

Gentlest Ascent Dynamics

W. E, X. Zhou, *Nonlinearity* **24**, 1831 (2011); J. M. Bofill, W. Quapp, M. Caballero, *Chem. Phys. Lett.* **583**, 203 (2013)

2.- a) Variational Nature of the Intrinsic Reaction Coordinate (Steepest Descent Path).

Variational Nature of Steepest Descent (Intrinsic Reaction Coordinate) Path. The Intrinsic Reaction Coordinate is the path that follows a gradient curve. It is also known as Steepest Descent (SD) or Ascent curve (SA). Introduced by: K. Fukui, J. Phys. Chem. **74**, 4161 (1970). Mathematically is characterized by: tangent curve $= \mathbf{t}(\mathbf{x}) = d\mathbf{x}/dt = \nabla_{\mathbf{x}}V(\mathbf{x})$



- On a PES there exists a field of gradients.

- Except for sationary points, at each point of the PES only a Steepest Descent or Steepest Ascent passes through this point.

- The Intrinic Reaction Coordinate is the only Steepest Descent or Steepest Ascent connecting two minimae through a first order saddle point (Transition State).

> The set of vectors indicates the Gradient field

A. Tachibana and K. Fukui, *Theor. Chim. Acta* **57**, 81 (1980); K. Fukui, *Int. J. Quantum Chem.*, *Quantum Chem. Symp.* **15**, 633 (1981).

R. Crehuet, J.M. Bofill, J. Chem. Phys. 122, 234105 (2005).

Using the Theory of Calculus of Variations.

$$I_{M \to q}\left(\mathbf{q}\right) = \int_{0}^{t} \sqrt{G(\mathbf{q})} \sqrt{\left(d\mathbf{q}/dt'\right)^{T} \left(d\mathbf{q}/dt'\right)} dt$$

Arc length: ds

We take $(\mathbf{q}_{M}, t' = 0)$ as fixed inital point and $(\mathbf{q}_{q'}, t' = t)$ as variable end point.

Gradient norm: $G(q)=(g^T(q)g(q))$. Speed Law. Where, g(q), is the gradient vector of the Potential Energy Surface.

The tangent of the path q(t) that extremalizes the above <u>variational</u> <u>integral</u>, $I_{M->q}(q)$, is: dq/dt = g(q). <u>Conclusion: the extremal curve is a</u> <u>Steepest Descent/ Ascent.</u> Evaluating the integral , $I_{M->q}(\mathbf{q})$ through the Steepest Ascent curve, we obtain, $I_{M->q}(\mathbf{q}) = V(\mathbf{q}_q) - V(\mathbf{q}_M)$.

At the point $(\mathbf{q}_q, t' = t)$: $dI_{M->q}(\mathbf{q})/dt = dV/dt = \mathbf{g}^T(\mathbf{q})d\mathbf{q}/dt$, a total differential form.

<u>Impact</u>: a Steepest Ascent or Steepest Descent curve, starts at the point \mathbf{q}_{M} , propagates through the PES according to the speed law or continuous slowness model, $(G(\mathbf{q}))^{1/2}$, arrives at the point \mathbf{q}_q , traveling with the extremal (least) potential energy variation, $I_{M->q}(\mathbf{q}) = V(\mathbf{q}_q) - V(\mathbf{q}_M)$. (Fermat Variational Principle).

Hamilton-Jacobi equation. At the variable end point $(\mathbf{q}_q, t' = t)$ we have $dI_{M-}(\mathbf{q}) = dV$ and from this we derive the **Eikonal equation**

Speed law
(Gradient Norm)
$$\xrightarrow{\left(\nabla_{\mathbf{q}}V\right)^{T}\left(\nabla_{\mathbf{q}}V\right)} = 1$$
 where $\nabla_{\mathbf{q}} = \begin{pmatrix} \partial/\partial q_{1} \\ \vdots \\ \partial/\partial q_{N} \end{pmatrix}$

Let $V(\mathbf{q})$ a solution of the above equation, then the Steepest Ascent/Descent curves (extremal curves) transverse the family of equipotential energy surfaces $V(\mathbf{q}) = v = constant$.

The construction of solutions of the eiconal equation as a set of equipotential energy surfaces is similar to the **Fermat–Huyghens principle** for the construction of wave fronts.

J.M. Bofill, W. Quapp, M. Caballero, J. Chem. Theory Comput. 8, 4856 (2012)



These equations are the basis of many algorithms to compute the Steepest Ascent/Descent (Intrinsic Reaction Coordinate) reaction path, see e.g. the review: H. B. Schlegel, *J. Comput. Chem.* **24**, 1514 (2003). One of the most widely used algorithm to integrate these curves: G. Henkelman, B. P. Uberuaga, and H. Jónsson, *J. Chem. Phys.* **113**, 9901 (2000).

<u>Second order variations</u>. We compare the value of the basic integral, $I_{M->q}(\mathbf{q})$, evaluated through an arbitrary curve (AC) and that evaluated through the Steepest Ascent curve both joining the same initial and final points, namely, $(\mathbf{q}_{M'}, \mathbf{t}' = 0)$ and $(\mathbf{q}_{q'}, \mathbf{t}' = \mathbf{t})$.

$$I_{M \to q} \left(\mathbf{q}_{AC} \right) - I_{M \to q} \left(\mathbf{q}_{SA} \right) =$$

$$\int_{0} \sqrt{G(\mathbf{q}_{AC})} \sqrt{\left(d\mathbf{q}_{AC}/dt'\right)^{T} \left(d\mathbf{q}_{AC}/dt'\right) dt'} -$$

 $\frac{dt'}{dt'} - \int_{0}^{t} \sqrt{G(\mathbf{q}_{SA})} \sqrt{(d\mathbf{q}_{SA}/dt')^{T} (d\mathbf{q}_{SA}/dt')} dt'$ the points Computed through the Steepest Ascent Curve (SA) that joints the points

Computed through the Arbitrary Curve (AC) that joints the points M and q.

This difference between integrals is the <u>Weierstrass E-function</u> or <u>Error Function</u>, that in the present case is **always positive**. The Weierstrass E-function is related with the second or higher order variation of the tangent argument, dq/dt. The Steepest Ascent (Descent) curves make positive the Error Function. Nevertheless, this is a necessary condition but not sufficient to ensure that any Steepest Ascent (Descent) curve minimizes the functional integral $I_{M>q}(q)$. The sufficient condition is satisfied if and only if the Steepest Ascent (Descent) curve joining two consecutive minimae does not contain a point that is second of higher order saddle point. These points are conjugate points of the starting minima. <u>The Intrinsic Reaction Coordiante Path is the</u> <u>unique Steepest Descent curve of character minimum.</u>

Computed through the Steepest Ascent Curve (SA) that joints the points M and q.

The second order variation: R. Crehuet, J.M. Bofill, *J. Chem. Phys.* **122**, 234105 (2005); A. Aguilar-Mogas, R. Crehuet, X. Giménez, J. M. Bofill, *Mol. Phys.* **105**, 2475 (2007); W. Quapp, *Theor. Chem. Acc.* **121**, 227 (2008).



Conjugate Point: If an extremal curve starting at any point and a second "neighboring" extremal curve, also starts at the same point, intersects to the first curve in a next point, then this intersection point is a conjugate point.

<u>Applications</u>: Location of an Intrinsic Reaction Coordinate curve between two minima using the minimization of the Weierstrass E-function:



Algorithm details: R. Crehuet, J.M. Bofill, *J. Chem. Phys.* **122**, 234105 (2005).



The **white open dots** are the set of 21 points of the initial guess curve. The **dark dots** indicate the final converged position of the 21 points. In this final position, all points are located in the <u>Intrinsic Reaction Coordinate curve</u>.

The **dark dots** are the set of 21 points of the <u>guess curve</u>. The point *R* is labeled as 1 and the point *P* as 21. The bold faced arrows are the gradient vectors of the **Weierstrass E-function** computed at each point of the guess curve.

2.- b) Variational Nature of Newton Trajectory (Distinguished Coordinate Path).

The other curve used as Reaction Path is the Distinguished or Driven Coordinate Path (DC), (M. J. Rothman and L. L. Lohr, Jr., Chem. Phys. Lett. **70**, 405 (1980)) or a more recent version, the so-called **Reduced Gradient Following** (RGF), (W. Quapp, M. Hirsch, O. Imig, D. Heidrich, J. Comput. Chem. **19**, 1087 (1998), J. M. Anglada, E. Besalú, J. M. Bofill, R. Crehuet, J. Comput. Chem. **22**, 387 (2001)) also labeled as **Newton Path** or **Newton Trajectory** (NT) (W. Quapp, M. Hirsch, D. Heidrich, Theor. Chem. Acc. **100**, 285 (1998)).

The **Reduced Gradient Following** or **Newton Trajectory Reaction Path** is characterized by a curve in the PES such that at each point of this curve, the gradient vector points at a constant direction.

This can be seen in another way, the **Reduced Gradient Following** curve crosses the steepest descent curve at each point so that at the same point the tangent has the same direction as the constant direction of the prescribed **Reduced Gradient Following** direction.

The **Reduced Gradient Following** or **Newton Trajectory Reaction Path** possesses other important features largely studied by Hirsch and Quapp (M. Hirsch, W. Quapp, *J. Math. Chem.* **36**, 307 (2004)) in their studies on the convexity of the PES region where the reaction path is located.



The Müller-Brown potential energy surface, E(x,y). **Reduced Gradient Following** or **Newton Trajectory** solutions $(E_x = 0, E_y)$ (dashed curve) and $(E_x, E_y = 0)$ (bold curve). They connect the three minima with the two saddle points. TP marks one of the turning points of the **Reduced Gradient Following** or **Newton Trajectory** curve $(E_x, E_y = 0)$.

The concept may be generalized by the challenge that any selected gradient direction is fixed

$$\nabla_{\mathbf{x}} E(\mathbf{x}) / \| \nabla_{\mathbf{x}} E(\mathbf{x}) \| = \mathbf{r}$$

where **r** is the selected unit vector of the search direction.

W.Quapp, M.Hirsch, O.Imig, D.Heidrich, J.Comput.Chem. 19, 1087 (1998).

The **Reduced Gradient Following** approach or **Driven Coordinate** method shows an important analogy to the mathematical theory of Branin, the **global Newton method**, that is given by the equation:

$$\frac{d\mathbf{x}}{dt} = \pm \mathbf{A}(\mathbf{x})\mathbf{g}(\mathbf{x})$$
 equivalent behavior to

$$\frac{\mathbf{g}(\mathbf{x})}{|\mathbf{g}(\mathbf{x})|} = \mathbf{r}.$$

Where A(x) is the adjoint matrix of the Hessian matrix H(x) and g(x) is the gradient vector all computed at the position x, r is the predefined unit vector.

The ``+" option is used for searching stationary points with odd index (stationary points with an odd number of negative eigenvalues of the Hessian), where the ``-" option searches for stationary points with even index (minima, or stationary points with an even number of negative eigenvalues of the Hessian). (W.Quapp, M.Hirsch, D.Heidrich, *Theor. Chem. Acc.* **100**, 285 (1998)).

Limit points: stationary points (g(x) = 0) and points where $g(x) \neq 0$ but A(x)g(x) = 0. In these points the **Newton Trajectory** or **Reduced Gradient Following** curve bifurcates.

The **Reduced Gradient Following** curves or that is the same the **Newton Trajectory** curves are <u>extremal curves of the integral functional</u> (J. M. Bofill, W. Quapp, *J. Chem. Phys.* **134**, 074101 (2011))

$$I\left(\mathbf{\bar{x}}\right) = \int_{x_{rc}^{0}}^{x_{rc}} V\left(x_{rc}, \mathbf{\bar{x}}\right) dx_{rc}^{'} \text{ where } \mathbf{\bar{x}}^{T} = \left(x_{1}, \cdots, x_{rc-1}, x_{rc+1}, \cdots, x_{N}\right)$$

and $V(\mathbf{x})$ is the potential energy surface, $\mathbf{x}^{T} = \left(x_{rc}, \mathbf{\bar{x}}^{T}\right)$.
The Euler-Lagrange equation is
 $\nabla_{\mathbf{\bar{x}}} V\left(x_{rc}, \mathbf{\bar{x}}\right) = \mathbf{0}_{N-1}$ because $\partial V/\partial x_{rc} \neq 0$ then $\nabla_{\mathbf{x}} V\left(x_{rc}, \mathbf{\bar{x}}\right) / \left\| \nabla_{\mathbf{x}} V\left(x_{rc}, \mathbf{\bar{x}}\right) \right\| = \mathbf{r} = \begin{bmatrix} 0_{1} \\ \vdots \\ 0_{rc-1} \\ 1_{rc} \\ 0_{rc+1} \\ \vdots \\ 0_{rc+1} \end{bmatrix}$
This set of equations determines the reaction path function $\mathbf{x} = \mathbf{x}(x_{rc}) \begin{bmatrix} 0_{1} \\ \vdots \\ 0_{rc+1} \\ \vdots \\ 0_{N} \end{bmatrix}$

<u>The tangent of this curve is:</u> $d\mathbf{x} / dx_{rc} = \mathbf{A}(\mathbf{x})\mathbf{g}(\mathbf{x}) = \mathbf{A}(\mathbf{x})\nabla_{\mathbf{x}}V(\mathbf{x}).$

Where the A(x) matrix is the adjoint of the Hessian matrix.

The extremal curve $\mathbf{x} = \mathbf{x}(x_{rc})$ makes the integral $I(\mathbf{x}) = \int_{x_{rc}^0}^{x_{rc}} V(x_{rc}, \mathbf{x}) dx_{rc}$ a minimum: <u>If the determinant of the Hessian matrix</u>, $\nabla_{\mathbf{x}} \nabla_{\mathbf{x}}^T V(\mathbf{x})$ <u>projected in the subspace orthogonal to r is positive definited at each point</u> <u>of the curve</u>, det $[\mathbf{S}^T(\nabla_{\mathbf{x}} \nabla_{\mathbf{x}}^T V(\mathbf{x}))\mathbf{S}]_{\mathbf{x}=\mathbf{x}(x_{rc})} > 0;$ $\mathbf{S} = [\mathbf{s}_1 \cdots \mathbf{s}_{N-1}];$ $\mathbf{S}^T \mathbf{r} = \mathbf{0}_{N-1}.$

V(x,y): <u>The Wolfe-Quapp Potential Energy Surface</u>.





In the turning point the **Newton Trajectory** curve has $dV/dx_{rc} = 0$

At each point of the valley-ridge border line the determinant of the Hessian matrix projected in the subspace orthogonal to the gradient is zero. The matrix **S** collects the set of linear independent vectors orthogonal to g(x).



A **Newton Trajectory** curve starting in a minimum minimizes the variational integral *I*(**x**) if the curve does not have a Valley Ridged Inflection point otherwise no statement can be made. In the former situtation the Newton Trajectory locates a Transition State and the whole curve is located in a walley. The Newton Trajectory is a Reaction Path with character Minimum Energy Path. (J. M. Bofill, W. Quapp, *J. Chem. Phys.* **134**, 074101 (2011)).

Newton Trajectories $(d\mathbf{x}/dt = \mathbf{A}(\mathbf{x})\mathbf{g}(\mathbf{x}))$ locate both stationary points, $\mathbf{g}(\mathbf{x}) = \mathbf{0}$ and Valley-Ridged-Inflection points $\mathbf{A}(\mathbf{x})\mathbf{g}(\mathbf{x}) = \mathbf{0}$. It can be used to locate Conical Intersections. (W. Quapp, J. M. Bofill, M. Caballero, *Chem. Phys. Lett.* **541**, 122 (2012)).

The **Reduced Gradient Following** or **Newton Trajectory** curves open a cornucopia of insights into the structure/topography of the Potential Eenergy Surface. To get this we need to test a somehow greater number of **Newton Trajectory** curves.

- For applications see e.g., the topography of cyclopropyl radical (W. Quapp, J.M. Bofill, J. Aguilar-Mogas, *Theor. Chem. Acc.* **129**, 803 (2011); W. Quapp, J.M. Bofill, *J. Math. Chem.* **50**, 2061 (2012)); ring opening cyclobutene (W. Quapp, J.M. Bofill, *Int. J. Quantum Chem.* **115**, 1635 (2015)).

- Other applications to chemical reactivity:

M. Hirsch, W. Quapp, J. Mol. Struct. THEOCHEM 683, 1 (2004).

- Newton Trajectories as basis of a Theory for Mechanochemistry and Catalysis:

W. Quapp, J.M. Bofill, J. Phys. Chem. B **120**, 2644 (2016); W. Quapp, J.M. Bofill, *Theor. Chem. Acc.* **135**, 113 (2016); W. Quapp, J.M. Bofill, J. Bofill, J. Comput. Chem. xx, yyy, (2016); W. Quapp, J.M. Bofill, J. Ribas-Ariño, Phys. Chem. Chem. Phys. (submitted). Also the lecture of Dr. W. Quapp in this meeting.

The curve where at each point the gradient norm is stationary in the equipotential surface is called Gradient Extremals.



J. Pancir, Collect. Czech. Chem. Commun. 40, 1112 (1975); M. V. Basilevsky, A. G. Shamov, Chem. Phys. 60, 347 (1981); M. V. Basilevsky, Chem. Phys. 67, 337 (1982); D. K. Hoffman, R. S. Nord, K. Ruedenberg, Theor. Chim. Acta 69, 265 (1986); P. Jørgensen, H. J. Aa. Jensen, T. Helgaker, Theor. Chim. Acta 73, 55 (1988); W. Quapp, Theor. Chim. Acta 75, 447 (1989); H. B. Schlegel, Theor. Chim. Acta 83, 15 (1991); J.-Q. Sun, K. Ruedenberg, J. Chem. Phys. 98, 9707 (1993); K. Bondensgard, F. Jensen, J.Chem.Phys. 104, 8025 (1996).

This curve is not widely used due to their computation cost. **Nevertheless, still is very appropriated as Reaction Path and in the Theory of the Mechanochemistry** (W. Quapp, J. M. Bofill, *Theor. Chem. Acc.* **135**, 113 (2016) Dr. W. Quapp in this meeting).



There does not exist an Intrinsic Reaction Coordinate Path (the most videly used type of path) connecting the minimum A with the minimum C.





Classical dynamics.

Quantum dynamics.



Mathematical Basis: An example of Lagrange-Bolza Variational Problem.

$$I(\mathbf{q}) = \int_{t_0}^{t'} \left\{ \frac{1}{2} \mathbf{g}^T(\mathbf{q}) \mathbf{g}(\mathbf{q}) - \lambda(\mathbf{q}) \left(V(\mathbf{q}) - \nu \right) \right\} dt$$

The resulting Euler-Lagrange equation: $\mathbf{Hg} = \lambda \mathbf{g}$ where **H** is the Hessian matrix and **g** is the gradient vector at the point **q**. At each point of a Gradient Extremal curve this eigenvalue equation is satisfied.

The eigenvalue equation, $H(q)g(q) = \lambda(q)g(q)$, determines the Gradient Extremal (GE) curve, q = q(t), implicitly. We note that in this case the boundary values, $q_0 = q(t_0)$ and $q_f = q(t_f)$, cannot be prescribed arbitrarily if the problem should have a solution. The tangent curve was first derived by J.-Q. Sun, K. Ruedenberg, J. Chem. Phys. **98**, 9707 (1993):



Using the perturbation theory applied to the eigenvalue equation, Hg = λ g, as formulated by McWeeny (R. McWeeny, *Phys. Rev.* **126**, 1028 (1961)) can be derived this tangent curve equation (J. M. Bofill, W. Quapp, M. Caballero, *J. Chem. Theory Comput.* **8**, 927 (2012)).

Special points of the Gradient Extremals (GE) curve.

$$\left(\mathbf{I} - \frac{\mathbf{g}\mathbf{g}^{T}}{\mathbf{g}^{T}\mathbf{g}}\right) \left[\left\langle \mathbf{F}\mathbf{g} \right\rangle + \mathbf{H}^{2} - \frac{\mathbf{g}^{T}\mathbf{H}\mathbf{g}}{\mathbf{g}^{T}\mathbf{g}} \mathbf{H} \right] \frac{d\mathbf{q}}{dt} = \mathbf{0}$$



The points where the matrix in the brakets is not invertible are **Turning Points** or **Bifurcation Points** of the Gradient Extremals curve.

Bifurcation Point

K. Bondensgård, F. Jensen, J. Chem. Phys. 104, 8025 (1996); J. M. Bofill, W. Quapp, and M. Caballero, J. Chem. Theory *Comput.* **8**, 927 (2012)

Turning Point

GE: Gradient Extremals curve

The Extremal Sufficient Conditions. Conditioned by the existence of Conjugate Points in the Gradient Extremal Curves.

<u>Necessary condition</u>, the stationary condition: $Hg = \lambda g$.

<u>Sufficient condition</u>, the second variation: $\delta^2 I(\mathbf{q}) > 0$ minimum ($\delta^2 I(\mathbf{q}) < 0$ maximum).

$$\delta^2 I(\mathbf{q}, \mathbf{p}) = \int_{t_0}^{t'} \mathbf{p}^T \left[\langle \mathbf{F} \mathbf{g} \rangle + \mathbf{H}^2 - \lambda \mathbf{H} \right] \mathbf{p} \, dt = \int_{t_0}^{t'} \mathbf{p}^T \mathbf{C} \mathbf{p} \, dt$$

Where $\mathbf{p}^T \mathbf{g} = 0$ and $\mathbf{p}(t_0) = \mathbf{0}$. If det (**C**) > 0 along the interval $t_0 \le t \le t'$ then the **Gradient Extremals** curve minimizes the variational integral, $I(\mathbf{q})$, otherwise maximizes $I(\mathbf{q})$.

If the **Gradient Extremals** curve from t_0 to t_1 the det (**C**) > 0, but at t_1 the det (**C**) = 0 and from this point until t' the det (**C**) < 0 then the **Gradient Extremals** curve loses the minimum character. There exists an arbitrary curve joining the the same points $q(t_0)$ and q(t') that makes the value of the integral I(q) lower with respect to the value of the same integral computed using the **Gradient Extremals** curve.

The points where the **Gradient Extremals** curve have det (C) = 0 are Turning Points or Bifurcation Points. The **Turning Points** and **Bifurcation Points** can be seen as the **Conjugate Points of Gradient Extremals curves**.



Basic Gradient Extremals Variational Integral: $I(\mathbf{q}) = \int_{t_0}^{t'} \{1/2\mathbf{g}^T(\mathbf{q})\mathbf{g}(\mathbf{q}) - \lambda(\mathbf{q})(V(\mathbf{q}) - \nu)\} dt$

I(**q**) evaluated on <u>GE curve (1)</u> > *I*(**q**) evaluated on <u>GE curve (2)</u>

We assume that the system under consideration is defined by a PES, $V(\mathbf{x})$, N is the number of degrees of freedom. The PES is assumed to be smooth. The $\mathbf{g}(\mathbf{x})$ is the gradient vector and $\mathbf{H}(\mathbf{x})$ the Hessian matrix at the point \mathbf{x} of $V(\mathbf{x})$. The equations that govern the **gentlest ascent dynamics** path (GAD) are as follows:

$$d\mathbf{x} / dt = -[\mathbf{I} - 2\mathbf{w}\mathbf{w}^T]\mathbf{g}(\mathbf{x}) = -[\mathbf{I} - \mathbf{w}\mathbf{w}^T]\mathbf{g}(\mathbf{x}) + \mathbf{w}\mathbf{w}^T\mathbf{g}(\mathbf{x}),$$

$$d\mathbf{w} / dt = -[\mathbf{I} - \mathbf{w}\mathbf{w}^T]\mathbf{H}(\mathbf{x})\mathbf{w} = -\mathbf{H}(\mathbf{x})\mathbf{w} + \mathbf{w}\mathbf{w}^T\mathbf{H}(\mathbf{x})\mathbf{w}, \qquad \mathbf{w}^T\mathbf{w} = 1.$$

Where I is the unit matrix.

<u>The first equation</u> means that we reverse the components of the gradient vector in the subspace orthogonal to $\mathbf{w}(t)$ and is preserved in the direction of $\mathbf{w}(t)$. <u>Thus it walks</u> towards a saddle point of index-one. The $[\mathbf{I} - 2\mathbf{w}\mathbf{w}^T]$ matrix is the orthogonal Householder transformation, $[\mathbf{I} - 2\mathbf{w}\mathbf{w}^T][\mathbf{I} - 2\mathbf{w}\mathbf{w}^T] = \mathbf{I}$ since $\mathbf{w}^T\mathbf{w} = 1$.

<u>The second equation</u> defines the dynamics of the ascent direction $\mathbf{w}(t)$. The first term on the right hand side ensures that $\mathbf{w}(t)$ converges to an eigenvector associated with the smallest eigenvalue of $\mathbf{H}(\mathbf{x})$ matrix. The second term ensures that the length of \mathbf{w} is fixed at 1. Note that $\mathbf{w}^{T}(t_{0})\mathbf{w}(t_{0}) = 1$.

GAD curve model was proposed by E and Zhou (W. E, X. Zhou, *Nonlinearity* **24**, 1831 (2011).)



Behaviour of the GAD curve on the Wolfe-Quapp PES. The GAD curve starts at the point near the minimum **Min**. The curve evolution ends at the **TS** point. The curve achieves the highest energy at the point **TP** (turning point). The curve leaves the valley where the starting minimum is located. The set of **w**-vectors generated during the search are indicated by the set of bold arrows.

<u>The turning point (TP)</u> occurs at the point of the GAD curve where the gradient vector, g(x), and the w-vector forms an angle equal to $\pi/4$ radians. (J.M. Bofill, W. Quapp, M. Caballero, *Chem. Phys. Lett.* **583**, 203 (2013))

The GAD curve is the counterpart of well-known classical navigation problem posed and solved by Zermelo an optimal control problem (E. Zermelo, Z. Angew. Math. Mech. 11, 114 (1931); C. Carathéodory, Variationsrechnung und partielle Differentialgleichungen erster Ordnung. B. G. Teubner, Berlin (1935)):

<u>Classical navigation problem</u>: it is given the present location of a ship in the sea, with a given current distribution characterized by a location dependent vector field. One desires to find the optimal control of the ship so as to reach the destination in the shortest possible time.

<u>Back to GAD curve</u>: the gradient vector field of the PES function can be thought of as representing the current of the sea, which we cannot change, whereas the normalized vector w determines the control. The destination is the next Stationary Point of the PES.

In a more precise way, the GAD model consists of the determination of the minimum of the t-parameter, $J[\mathbf{x}_{TS}(\mathbf{w}(t_f))] = t_f - t_0$, in which a controlled point can be evolved from a given minimum point of the PES, $\mathbf{x}_{Min} = \mathbf{x}(t_0)$, to a final transition state of this PES, $\mathbf{x}_{TS} = \mathbf{x}(t_f)$, boundary conditions, $\mathbf{g}(\mathbf{x}(t_0)) = \mathbf{g}(\mathbf{x}(t_f)) = \mathbf{0}$. The evolution of the test point is described by the system of ordinary differential equations: $d\mathbf{x}/dt = -[\mathbf{I} - 2\mathbf{w}\mathbf{w}^T]\mathbf{g}(\mathbf{x})$, where $\mathbf{w}(t)$ is a normalized N-dimensional vector of the control parameters.

Since t_0 is fixed, the required minimum t_f is merely the minimization of the functional J $[\mathbf{x}_{TS}(\mathbf{w}(t_f))]$ that depends on the chosen $\mathbf{w}(t)$ -control normalized vector. Thus the GAD curve model is a <u>t</u>-parameter-optimal control problem and can be considered as a particular instance of the Mayer problem of the Theory of Calculus of Variations. Also the GAD, as a case of an optimal control problem, must satisfy the Pontryagin Maximum Principle. (J.M. Bofill, W. Quapp, Theor. Chem. Acc. 135, 11 (2016); J.M. Bofill, W. Quapp, An Application of the Maximum Principle in Chemistry: A Method to Locate Transition States, Apple Academic Press, New Jersey (2016)).

Applying a device due to Zermelo (see also Carathéodory), the GAD Hamiltonian of this type Mayer variational problem is:

$$2H(\mathbf{x},\mathbf{y}) = (2\mathbf{w}^T \mathbf{g}(\mathbf{x}))^2 (\mathbf{y}^T \mathbf{y}) - (1 + \mathbf{y}^T \mathbf{g}(\mathbf{x}))^2 = 0 \quad \text{where} \quad \mathbf{w}(1 + \mathbf{y}^T \mathbf{g}(\mathbf{x})) = (2\mathbf{w}^T \mathbf{g}(\mathbf{x}))\mathbf{y}.$$

The two canonical equations associated to this Hamiltonian are the GAD system of ordinary differential equations, namely, $d\mathbf{x}/dt = -[\mathbf{I} - 2\mathbf{w}\mathbf{w}^{T}]\mathbf{g}(\mathbf{x})$ and $d\mathbf{w}/dt = -[\mathbf{I} - \mathbf{w}\mathbf{w}^{T}]\mathbf{H}(\mathbf{x})\mathbf{w}$, where $\mathbf{w}^{T}\mathbf{w}=1$. The minimum or maximum character of GAD extremal curve is analyzed through the positivity or negativity value of the Weierstrass error function, $E(\mathbf{x}, d\mathbf{x}/dt, d\mathbf{x'}/dt) = 2[1 - (\mathbf{w}^{T}\mathbf{w'})(\mathbf{g}^{T}(\mathbf{x})\mathbf{w'})/(\mathbf{g}^{T}(\mathbf{x})\mathbf{w})]$. This is an indirect proof of Pontryagin Maximum Principle. (J.M. Bofill, W. Quapp, Theor. Chem. Acc. **135**, 11 (2016)).



The turning point (**TP**) occurs at the point of the **GAD** curve where the gradient vector, $\mathbf{g}(\mathbf{x})$, and the $\mathbf{w}(t)$ -control vector forms an angle equal to $\pi/4$ radians.

A **GAD** curve (**blue**) by Eq. $d\mathbf{x}/dt = -[\mathbf{I}-2\mathbf{w}\mathbf{w}^T]\mathbf{g}$ on a two-dimensional toy potential. The control vector, w(t), is throughout the first eigenvector, calculated by Eq. $dw/dt = -[I-ww^T]Hw$. The surface is a modified Neria-Fischer-Karplus case [M. Hirsch, W. Quapp, Chem. Phys. Lett. **395**, 150 (2004)]. The asterisk (*) marks a quasi-shoulder, and the thin dashes mark the *borderline between valleys and ridges.* **TP** is the *turning point* of the **GAD** curve. For comparisons are given: the valley Gradient Extremal (GE) by a bold faced black curve, the *Intrinsic Reaction Coordinate* (IRC) by a red curve, and a Newton Trajectory (NT) by a dashed black curve. Note that the **IRC** starts near the **SP** of index one, but the **GAD** starts near the Min.

In this case GAD does not satisfy the Reaction Path requirements. It goes over the borderline between valley and ridge.

<u>Proposed Reaction Path curves based on the model of Zermelo's problem.</u> It is proposed a generic type curve where its general tangent is given by the expression:

$$d\mathbf{x} / dt = -\mathbf{g}(\mathbf{x}) + f(\phi, \mathbf{x}, \mathbf{w})\mathbf{w}, \qquad \mathbf{w}^T \mathbf{w} = 1.$$

Where **w** is the control vector. The function $f(\phi, \mathbf{x}, \mathbf{w})$ is a continuous and differentiable function with respect to **x** and ϕ is a constant. Taking $f(\phi, \mathbf{x}, \mathbf{w}) = \phi \mathbf{w}^T \mathbf{g}(\mathbf{x})$ with ϕ larger that one. The $d\mathbf{w}/dt$ is given by the expression:

$$d\mathbf{w} / dt = -\left[\mathbf{I} - \mathbf{w}\mathbf{w}^{T}\right] \left(\nabla_{\mathbf{x}} f\left(\phi, \mathbf{x}, \mathbf{w}\right) - \mathbf{H}\left(\mathbf{x}\right)\mathbf{w}\right) = -\left(\phi - 1\right) \left[\mathbf{I} - \mathbf{w}\mathbf{w}^{T}\right] \mathbf{H}\left(\mathbf{x}\right)\mathbf{w}.$$



Curves (blue with $\phi \ge 2$, red with $\phi = (2)^{1/2}$) to Eq. dx/ dt with $f(\phi, \mathbf{x}, \mathbf{w}) = \mathbf{g}^T \mathbf{w}$. Start is at minimum. The **bold blue** curve is the GAD curve with $\phi = 2$, the other blue curves are to $\phi = 3$, 4, and 10. The control vector is calculated by Eq. dw/dt . The asterisk (*) marks a quasi-shoulder, and the thin dashes mark the borderline between valleys and ridges. For comparison, the GE is given (thick black curve) which is here the valley floor pathway between **SP** and **Min**. (J.M.Bofill, W.Quapp, *Theor. Chem. Acc.* **135**, 11 (2016)).

<u>The extension of GAD to a kind of molecular dynamics.</u> It was proposed by Samanta and E (A. Samanta, W. E, J. Chem. Phys. **136**, 124104 (2012); J. M. Bofill, W. Quapp, E. Bernuz, J. Math. Chem. **53**, 41 (2015)). The dynamical equations are:

$$d\mathbf{x} / dt = \mathbf{p},$$

$$d\mathbf{p} / dt = -[\mathbf{I} - 2\mathbf{w}\mathbf{w}^{T}]\mathbf{g}(\mathbf{x}),$$

$$d\mathbf{w} / dt = -[\mathbf{I} - \mathbf{w}\mathbf{w}^{T}]\mathbf{H}(\mathbf{x})\mathbf{w}, \qquad \mathbf{w}^{T}\mathbf{w} = 1.$$

MD-GAD trajectory, depicted in red color, but starting at the point $(x, y, p_x, p_y, w_x, w_y)$ = (1.1,-1.4, 0, 0, 0, 1). The initial **w** vector is the second column of the unit matrix. The regions marked by a black square contain a Turning Point of the trajectory. The trajectory also shows 'some chaotic' behavior in the region where **TS1**, **TS3** and **MAX** stationary points are located.





An MD-GAD trajectory in red color, starting at the point (x, y, p_x , p_y , w_x , w_v) = (1.1,-1.4, 0, 0, 0.982,-0.189). The initial control vector, w, is the eigenvector of the second eigenvalue of the Hessian matrix evaluated at the starting point. The black arrows are the momentum or tangent vectors of the trajectory, dx/dt = p, the green arrows are the control w-vectors while the arrows in magenta are the gradient vectors of the PES, $\mathbf{g}(\mathbf{x})$. The regions marked by a black square are where a Turning Point of the trajectory occurs. This trajectory here does not show any 'chaotic' behavior.

(J. M. Bofill, W. Quapp, E. Bernuz, J. Math. Chem. 53, 41 (2015))

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